

Application of the WDXRF AR35 on-stream analyzer for determination of the content of Fe, Co, Ni, Cu and S in slurry products

INTRODUCTION

1.1 Determination of average elements in polymetallic ores

The analysis of the content of iron Fe, cobalt Co, nickel Ni and copper Cu during the technological process of polymetallic ore enrichment by X-ray fluorescence analysis



(XRF) using the analyzer AR-35 is a typical task and does not present any difficulties.

1.2 Determination of light elements in polymetallic ores



The determination of light elements by the XRF method is complicated by the fact that the fluorescent lines of these elements are strongly absorbed by air, as well as by the materials of

the separation windows all the way from the sample to the detector.

Direct analysis of sulfur S without additional improvements that reduce the absorption of lines of light elements is impossible due to the almost complete absorption of lines $SK_{\alpha,\beta}$ by the volume of air between the sample and the working surface of the detector.

EQUIPMENT AND METHODS

To measure the pulp samples, the AR-35 analyzer configuration was used, including:

- X-ray tube with Pd anode;
- spectrometric channels on the line FeK_{α} , CoK_{α} , NiK_{α} , CuK_{α} , SK_{α} and scattered radiation Pd_{inc} ;
- pneumatic shutter with primary X-ray filter;
- measuring cuvette with PET working and protective film.

Two measurement modes were used, which differ in the use of primary X-ray filtration.

TESTING ON REAL SLURRY SAMPLES

A set of 16 samples presented by four technological products was used as test objects:

- Condensed nickel Concentrate;



- Nickel-pyrrhotite concentrate;
- Metal-containing product;
- Condensed nickel sulfide concentrate.

Parameters of the constructed calibration dependencies:

Element	Content range, %	Regression equation	Residual calibration error σ, % масс.	Linear correlation coefficient, R
Fe	from 26.6 to 42.5	$C_{P}=a_0+a_1I_{FeK\alpha}+a_2I_{CuK\alpha}+a_3I_{Pdinc}$	0.503	1
Со	from 0.034 to 0.371	C _P = a ₀ + a ₁ _{C₀Kα} + a ₂ _{C₀Kα} + a ₃ _{FeKα}	0.0055	1
Ni	from 0.87 to 8.51	C _P = a ₀ + a ₁ I _{CoKα} + a ₂ I _{FeKα} + a ₃ I _{CuKα}	0.113	1
Cu	from 0.39 to 3.58	$C_{P}=a_0+a_1I_{CuK\alpha}+a_2I_{FeK\alpha}+a_3I_{Pdinc}$	0.058	1
S	from 17.1 to 30.1	C _P = a ₀ + a ₁ lsκ _α ·lsκ _α + a ₂ lsκ _α ·l _{Cuκα} + a ₃ lsκ _α ·l _{Feκα} + a ₄ l _{Pdinc}	0.393	1

The values of the detection limits calculated from measurements of real slurry samples:

Element	Fe	Со	Ni	Cu	S
Background intensity, imp/s	13.0	9.0	19.0	61.0	15.0
Cipher of the initial sample	11	22	33	44	55
Intensity of the line from the slurry sample, imp/s	8650	44.3	564	429	73
Concentration of the element in the sample, %	26.6	0.034	0.87	0.39	17.1
Exposition T, s	40	40	40	40	40
Definition limit, %	0.018	0.005	0.011	0.013	1.81



CONCLUSIONS

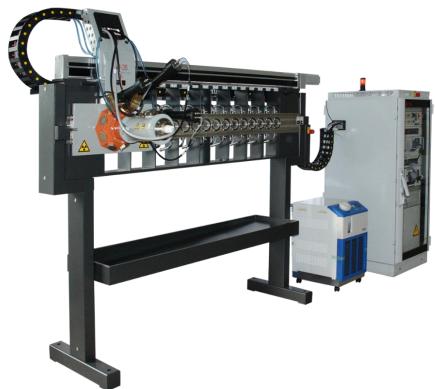
1. The on-stream WDXRF analyzer AR-35 makes it possible to determine the concentrations of iron Fe, cobalt Co, nickel Ni and copper Cu in the solid phase of liquid pulp samples with an intrinsic analysis error of up to 10% relative.

The determination of sulfur S can be carried out after the completion of the AR-35 analyzer and the installation of a spectrometric channel. The error of the sulfur S analysis will range from 5% to 20% relative, depending on the concentration range.

2. The values of the detection limits for Fe, Co, Ni and Cu are at the level of 0.01%, and for S the detection limit was 1.8%.

The value of the limits of determination and the final error of the analysis depends on the properties of the product: slurry density, fineness of grinding, mineralogical composition, etc. Methodological recommendations have been developed to improve the analysis indicators, which are selected individually for each enterprise.

3. The measurement time of one product starts from 40 seconds, the total cycle time for 15 lines is no more than 15 minutes.

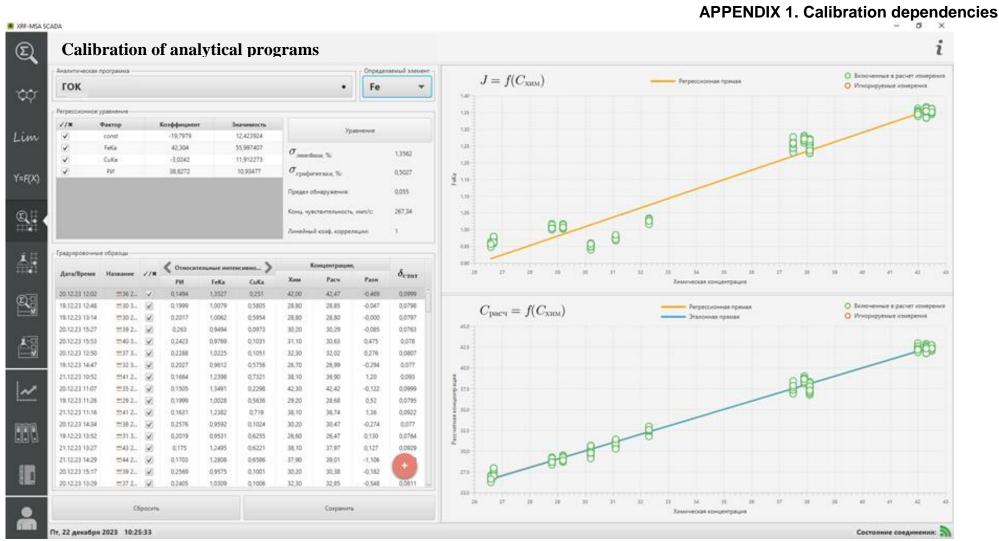


In the photo: AR-35 analyzer for 15 measuring cell, cooling system, automatic control stand

Applications:

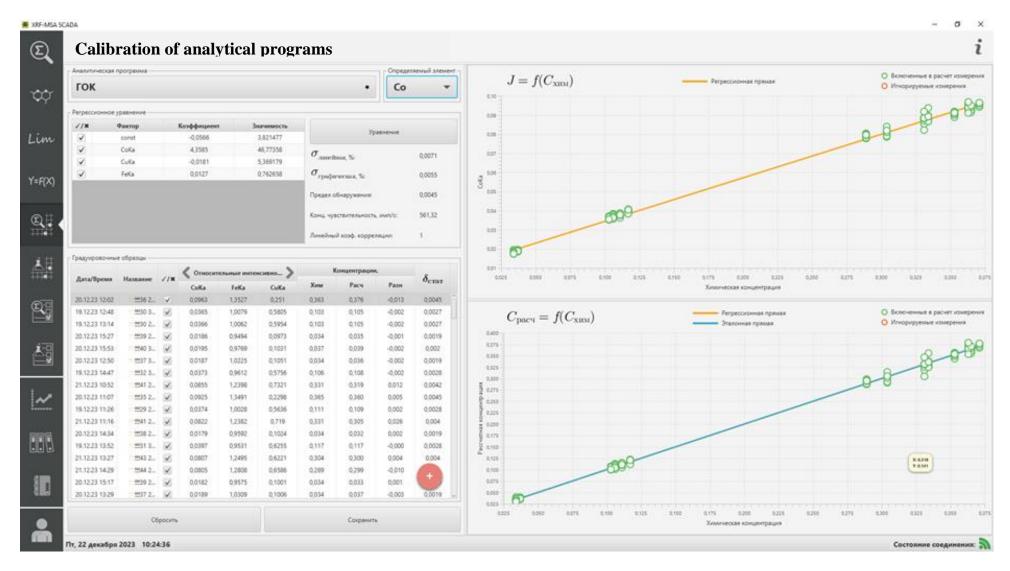
Appendix 1. Visualization of calibration.





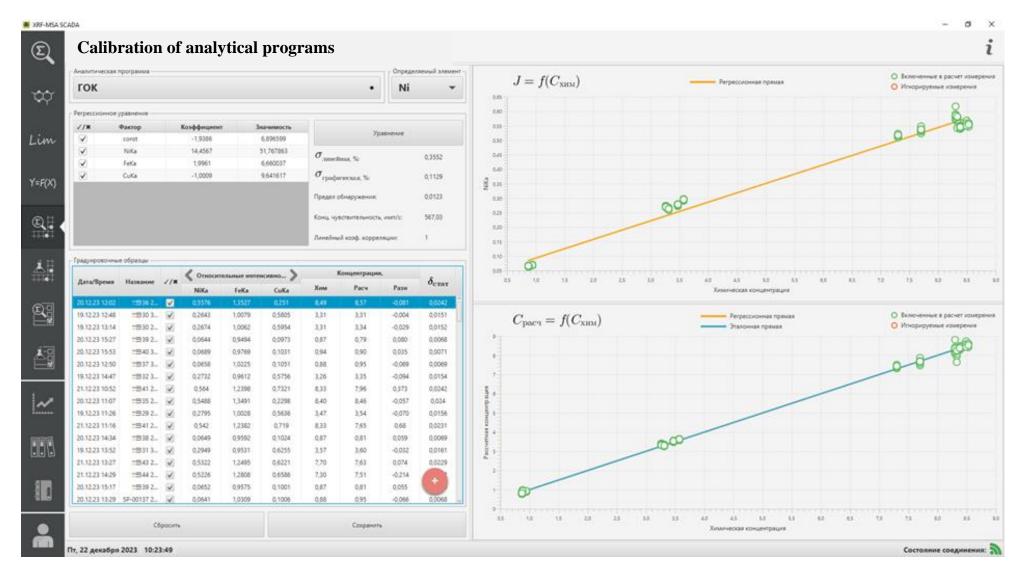
Picture 1 – Calibration dependence for Fe.





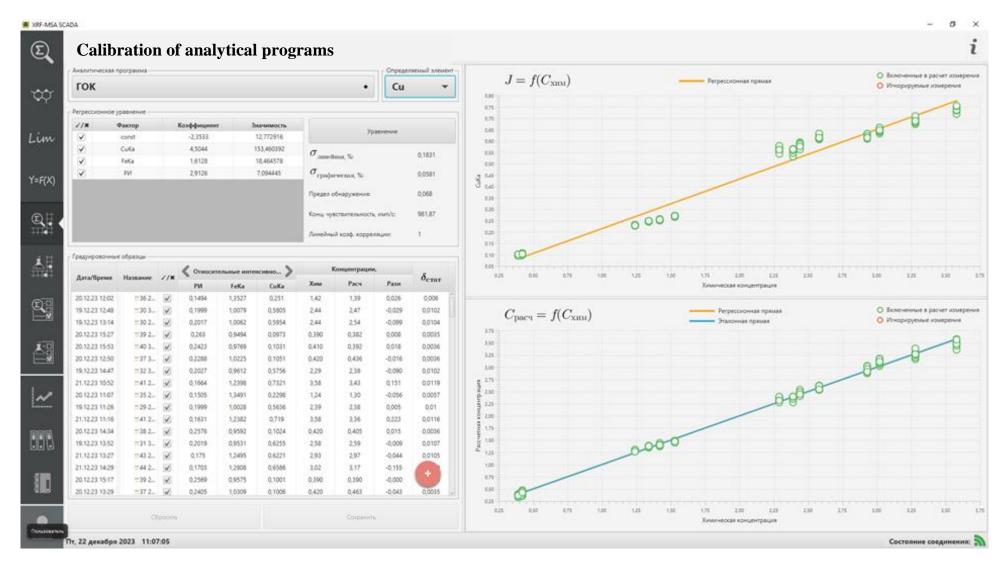
Picture 2 - Calibration dependence for Co.





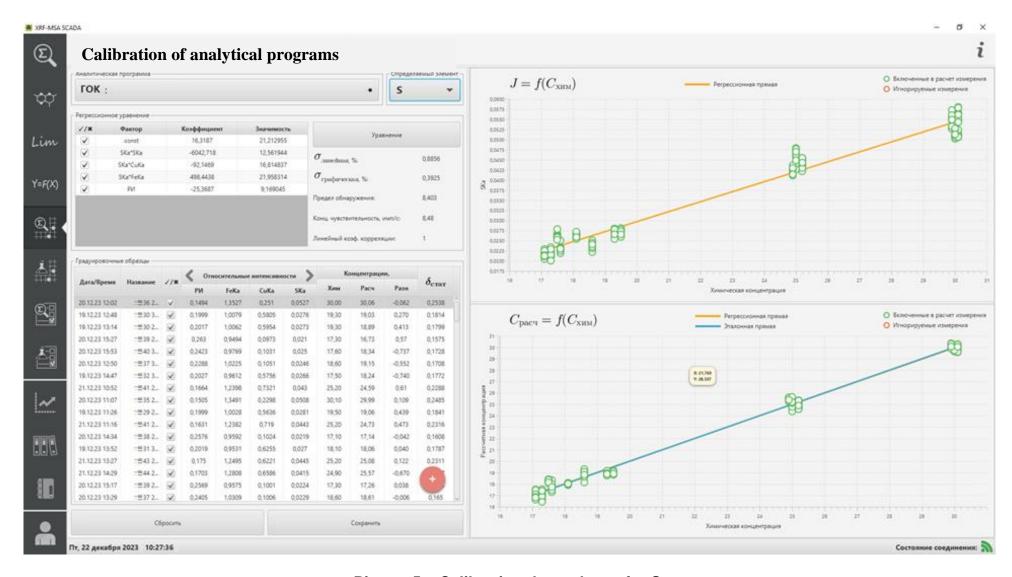
Picture 3 - Calibration dependence for Ni.





Picture 4 - Calibration dependence for Cu.





Picture 5 – Calibration dependence for S.